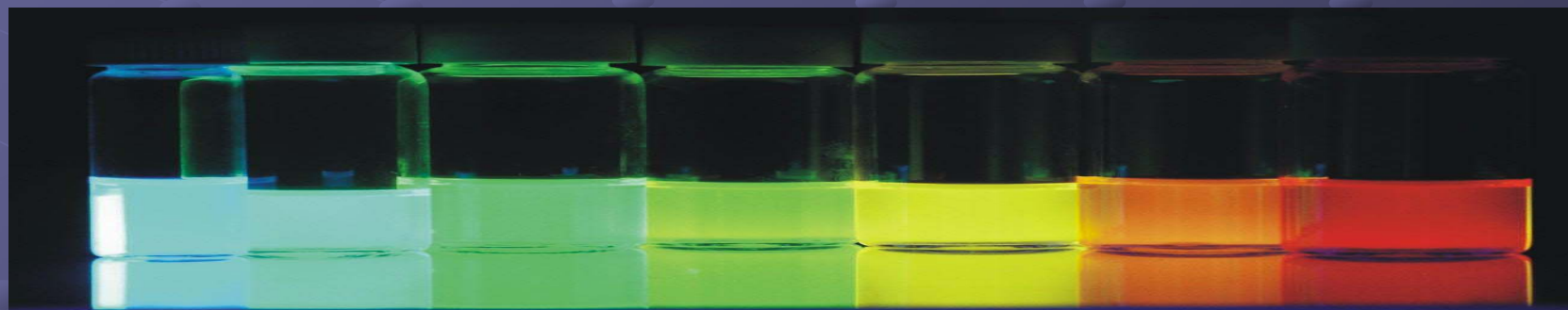


# Quantum Dots

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Size: 2.3nm

5nm

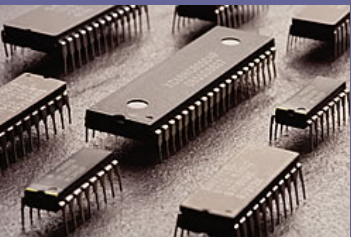
Source: [http://www.greenspine.ca/media/quantum\\_dots\\_c.jpg](http://www.greenspine.ca/media/quantum_dots_c.jpg)

# Why Quantum Dots?

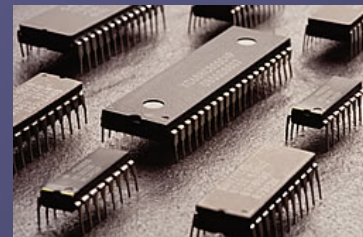
- I chose to research quantum dots because of an article that I read in a magazine over the summer. I thought that quantum dots sounded like they had a lot of applications and I wanted to know how they came up with the idea. I learned a lot about them and have come to the conclusion that they aren't as cool as I originally thought they were – **THEY'RE BETTER!**

# What are Quantum Dots?

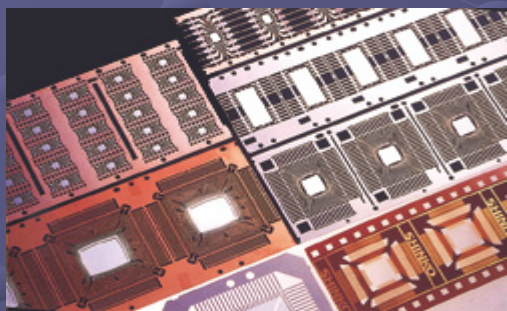
- Quantum dots are semiconductor nanocrystals that are so small they are considered dimensionless.
- Quantum dots range from 2-10 nanometers (10-50 atoms) in diameter.
- [Animation](#)



# Quick Review of Semiconductors



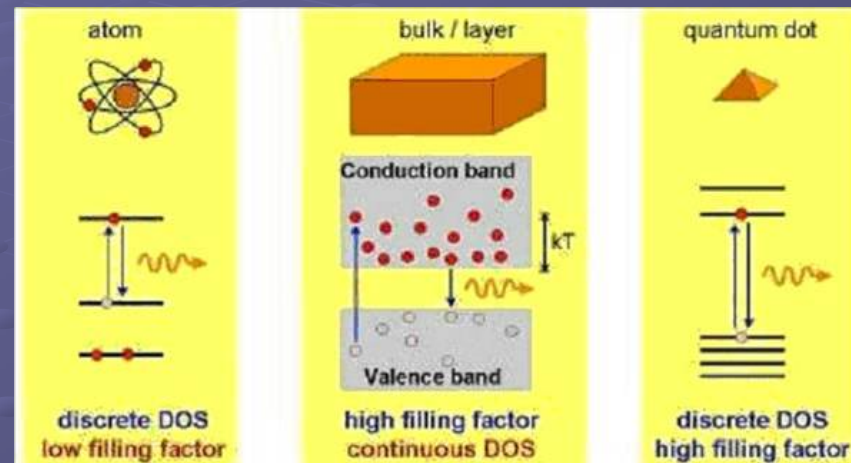
- A semiconductor is a material that has a small band gap between the valence and conduction band.
- An exciton pair is defined as an electron and the hole that it leaves behind when it is excited up to the conduction band.
- An exciton bohr radius is the distance in an electron-hole pair.
- A Quantum Dot is a semiconductor so small that the size of the crystal is on the same order as the size of the exciton bohr radius.
- This unique size property causes the “band” of energies to turn into discrete energy levels.



Source: <http://www.smtresource.com/web/images/sections/icon-semi-150x150.jpg>

<http://www.export-japan.com/hocci/worldmetal/img/worldmetal01.jpg> [http://www.cpu-world.com/CPU/32081/L\\_NS-NS32081D-6.jpg](http://www.cpu-world.com/CPU/32081/L_NS-NS32081D-6.jpg)

# Artificial Atoms?

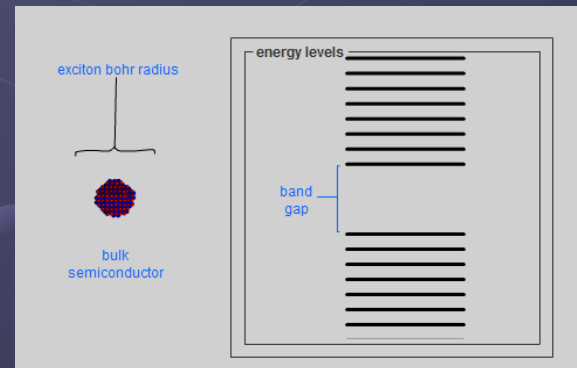
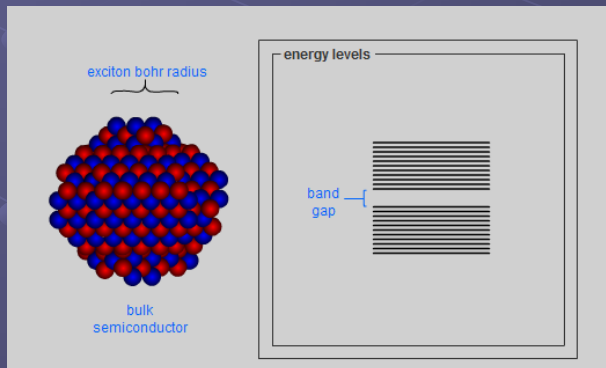


- Quantum Dots are more closely related to atoms than a bulk material because of their discrete, quantized energy levels.
- They have been nicknamed Artificial Atoms.



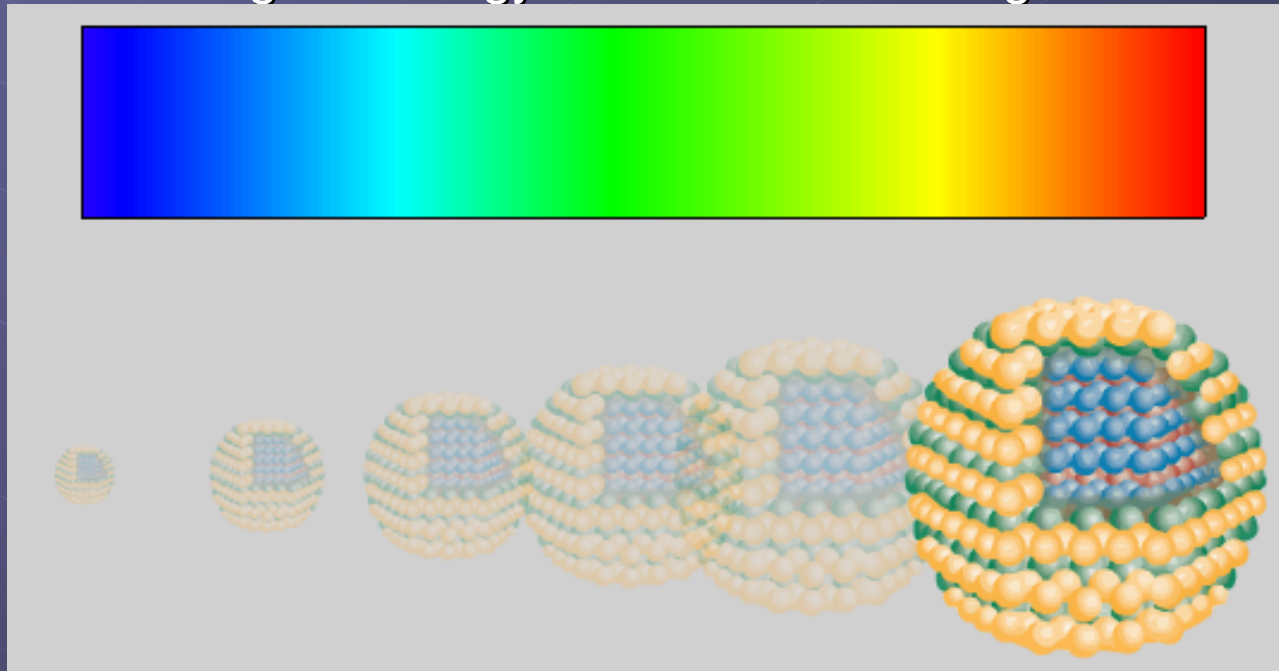
# Exciton Bohr Radius & The Band Gap

- Band Gaps, and therefore the energy associated with them, depend on the relationship between the size of the crystal and the exciton bohr radius.
- Animation



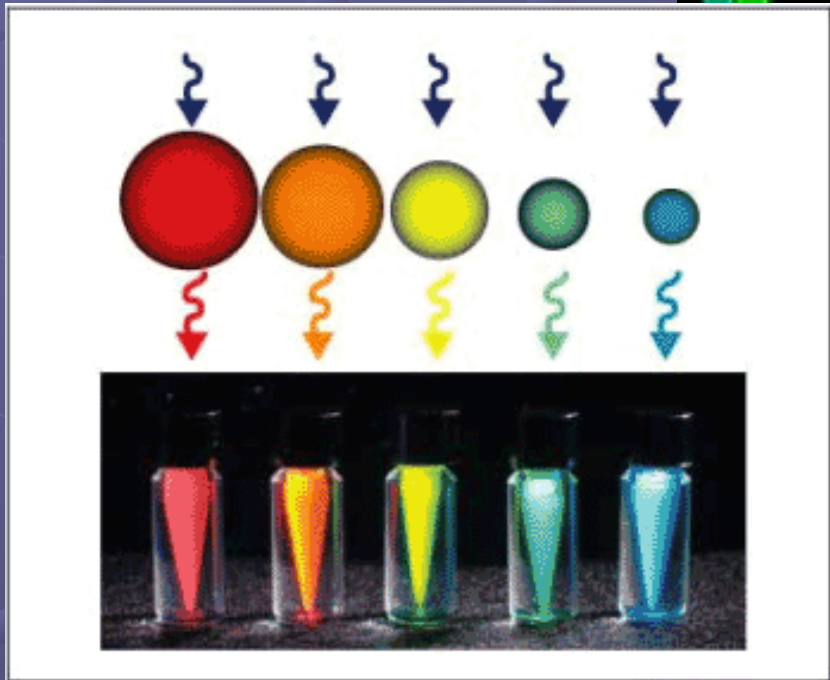
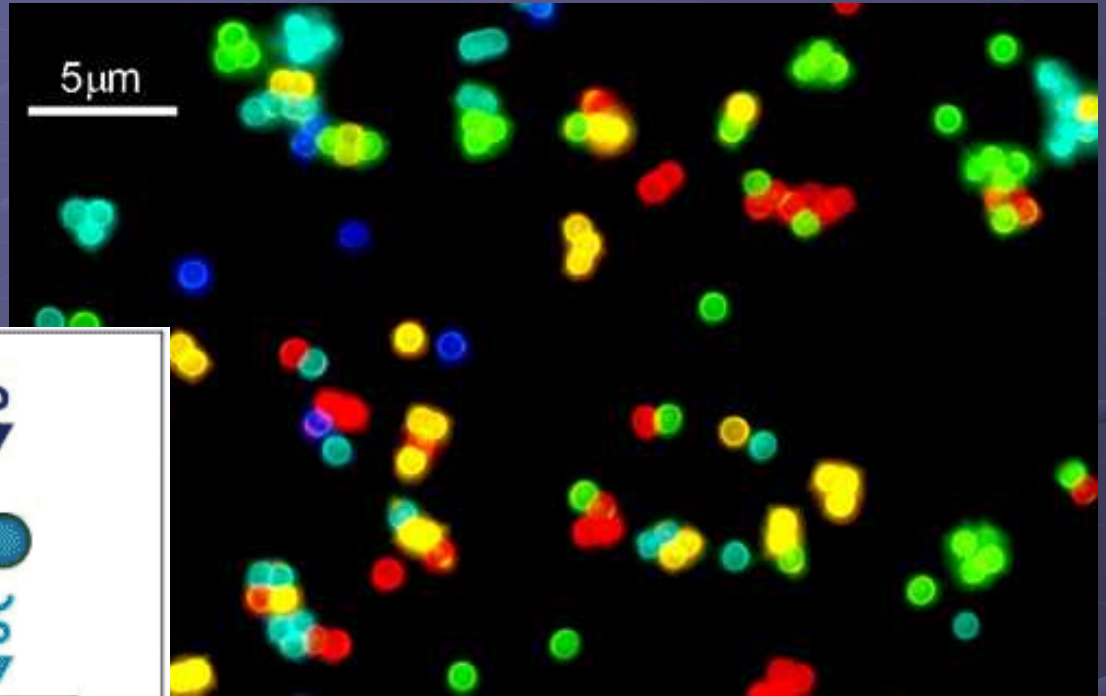
# Colorific Properties

- The height, and energy difference, between energy levels increases as the size of the quantum dot decreases (remember the animation we just saw).
- Smaller Dot = Higher Energy = Smaller Wavelength = Blue Color



Source: [http://www.evidenttech.com/qdot-definition/flash-player.php?mov=08\\_tunable](http://www.evidenttech.com/qdot-definition/flash-player.php?mov=08_tunable)

# Color and Quantum Dots



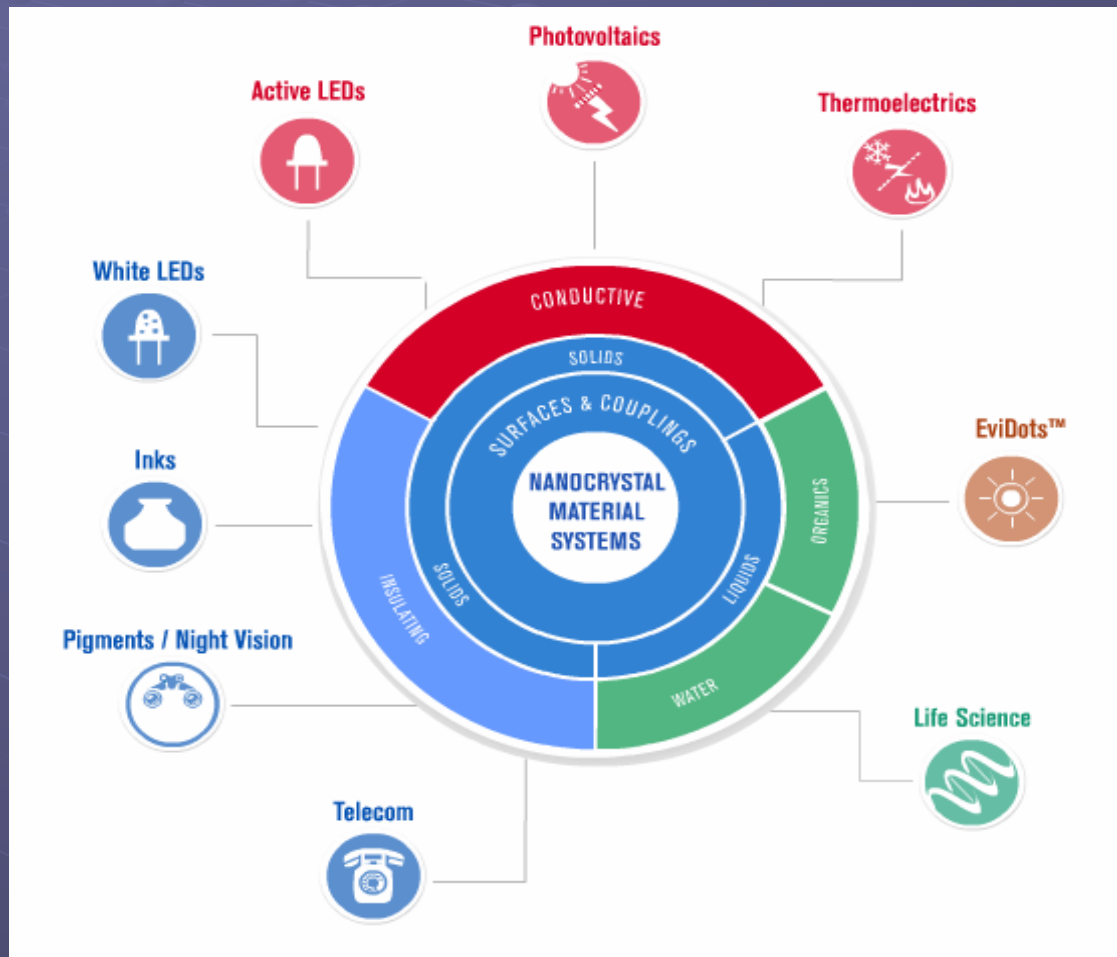
Source:

<http://optics.org/objects/news/7/7/2/Quantumdottag.jpg>

<http://www.qdots.com/live/images/figure9.gif>



# Applications of Quantum Dots

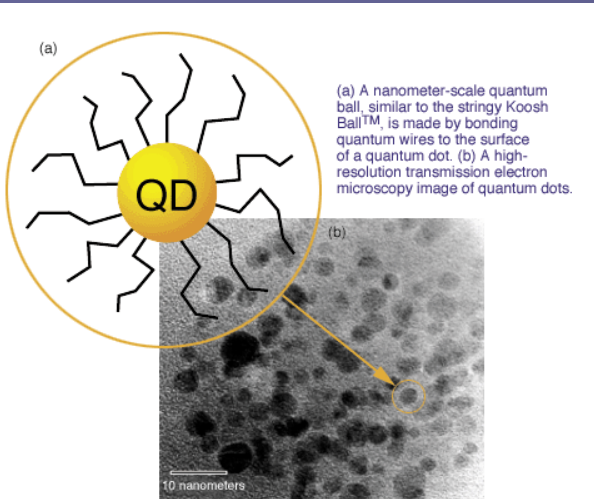


Source: <http://www.evidenttech.com/applications/quantum-dot-applications.php>

# Future Applications of Quantum Dots

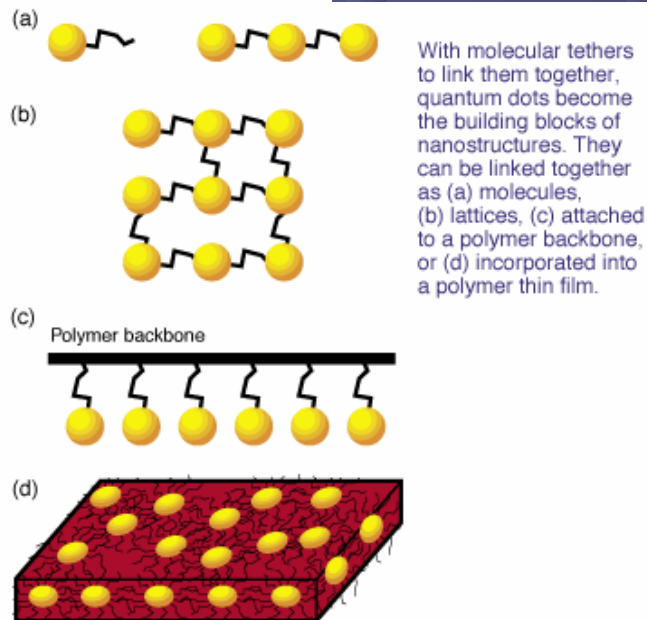
- White LED's – Replace Incandescent Light bulbs with red, green, and blue QD's to get white light.
- Active LED's – can get any color needed virtually pain free; since QD's are so small, they can be inserted into basically anything.
- Life Sciences – Can be used in place of traditional organic dyes; last longer and can be tuned more specifically than organic dyes.

# Applications - Circuits



By adding “quantum wires” to a quantum dot, many quantum dots have been strung together in various shapes and structures.

Hypothetically, these quantum dot/wire structures could be used in circuits in place of traditional wires.



# Conclusions

- Quantum Dots are a new and innovative perspective on the traditional semiconductor.
- Quantum Dots can be synthesized to be essentially any size, and therefore, produce essentially any wavelength of light.
- There are many possible applications of Quantum Dots in many different areas of industry/science.
- The future looks bright and exciting on all the possible applications of Quantum Dots.

# Acknowledgements

- I would like to thank the Evident Technology company and, more specifically, their IT people for the great animations they have created.